# Transition-Metal-Free Carbofluorination of TBS-Protected Nitrogen-Containing Cyclic Enynols: Synthesis of Fluorinated Azabicycles 

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ABTRACT: The synthesis of fluorinated azabicycles from tertbutyldimethylsilyl protected N -containing cyclic enynols using inexpensive $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ is described. In this reaction, $\mathrm{BF}_{3}$ reacts as both the Lewis acid and the fluoride source for cyclization/fluorination of the TBS-protected cyclic N containing enynols. The method provides an easy access to fluorinated azabicycles where a new $\mathrm{C}\left(\mathrm{sp}^{2}\right)-\mathrm{F}$ bond and a new bicyclic skeleton are generated at ambient temperature within 1-13 min under metal-free reaction conditions.


Scheme 1. Synthesis of Ketones 3a-d


Table 1. Optimizing of Reaction Conditions in the Carbofluorination of 1 a with $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$

|  |  |  | $\xrightarrow[\text { solvent }]{\text { Lewis acid }}$ |  <br> 2a |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |
| Entry | Lewis acid | Loading (equiv) | Solvent | Time | Yield $(\%)^{\mathrm{a}}$ |
| 1 | $\mathrm{BF}_{3} \mathrm{OEt}_{2}$ | 1 | $0.1 \mathrm{M} \mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 1 min | 25 |
| 2 | $\mathrm{BF}_{3} \mathrm{OEt}_{2}$ | 2 | $0.1 \mathrm{M} \mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 1 min | 48 |
| 3 | $\mathrm{BF}_{3} \mathrm{OEt}_{2}$ | 2 | 0.1 M DBE | 1 min | 34 |
| 4 | $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | 2 | 0.1 M DCE | 1 min | 33 |
| 5 | $\mathrm{BF}_{3} \mathrm{OEt}_{2}$ | 2 | $0.1 \mathrm{M} \mathrm{CHCl}_{3}$ | 1 min | 34 |
| 6 | $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | 2 | 0.1 M toluene | 1 min | 6 |
| 7 | $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | 2 | 0.01 M toluene | 15 min | 11 |
| 8 | $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | 2 | 2.5 mM toluene | 35 min | 25 |
| 9 | $\mathrm{BF}_{3} \mathrm{OEt}_{2}$ | 2 | $0.1 \mathrm{M} \mathrm{CH}_{3} \mathrm{CN}$ | 1 min | $0^{\text {b }}$ |
| 10 | $\mathrm{BF}_{3} \mathrm{OEt}_{2}$ | 10 | $0.1 \mathrm{M} \mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 1 min | 50 |
| 11 | $\mathrm{Ph}_{3} \mathrm{CBF}_{4}$ | 2 | $0.1 \mathrm{M} \mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 1 min | 26 |
| 12 | $\mathrm{Ph}_{3} \mathrm{CBF}_{4}$ | 5 | $0.1 \mathrm{M} \mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 1 min | 27 |
| 13 | $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | 2 | 0.1 M THF | 10 h | $0{ }^{\text {c }}$ |
| 14 | $n-\mathrm{Bu}_{4} \mathrm{NF}$ | 2 | 0.1 M THF | 0.5 h | $0^{\text {c }}$ |
| 15 | $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | 2 | 0.01 M THF | 36 h | $0^{\text {d }}$ |
| 16 | $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | 2 | $0.01 \mathrm{M} \mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 1 min | 51 |
| 17 | $\mathrm{BF}_{3} \mathrm{OEt}_{2}$ | 2 | $2.5 \mathrm{mM} \mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 1 min | 56 |

${ }^{\text {a }}$ Isolated yields by column chromatography. ${ }^{\text {b }}$ Compound 4 was isolated in $79 \%$ yield. ${ }^{\text {c }}$ Deprotection product 1a' was isolated in good yields. ${ }^{\text {d }}$ Compound 1a was recovered quantitatively.

Table 2. Substrate scope

solated diffraction analysis.

Scheme 3. Postulated Reaction Paths for Formation of 1a and 3a


1a
F


3a

