## Facile Synthesis of Azaspirocycles via Iron Trichloride-Promoted Cyclization/ Chlorination of Cyclic 8-Aryl-5-aza-5-tosyl-2-en-7-yn-1-ols

ORGANIC LETTERS

2012 Vol. 14, No. 7 1830–1833

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Table 1. Optimizing the Reaction Conditions

entry	Lewis acid	solvent	temp (°C)	time	yield (%)
1	FeCl <sub>3</sub>	$CH_2Cl_2$	23	1 min	83
2	$FeCl_3$	$\mathrm{CH_2Cl_2}$	0	$3 \min$	72
3	$AlCl_3$	$CH_2Cl_2$	23	1 min	$67^a$
4	$TiCl_4$	$CH_2Cl_2$	23	$0.5\mathrm{h}$	$17^b$
5	$\mathrm{SnCl}_4$	$CH_2Cl_2$	23	$0.5\mathrm{h}$	27
6	$ZnCl_2$	$CH_2Cl_2$	23	48 h	0
7	$Fe(NO_3)_3 \cdot 9H_2O$	$CH_2Cl_2$	23	48 h	0
8	$FeCl_3$	DCE	23	$1 \min$	74
9	$FeCl_3$	DBE	23	2 h	72
10	$FeCl_3$	THF	23	$26  \mathrm{h}$	23
11	$\mathrm{FeCl}_3$	$\mathrm{CH_{3}CN}$	23		0

 $^aZ/E = 6:1$  (determined by 400 MHz  $^1$ H NMR analysis of the crude reaction mixture).  $^bZ/E = 13:1$  (determined by 400 MHz  $^1$ H NMR analysis of the crude reaction mixture).

**Table 2**. FeCl<sub>3</sub>-Promoted Cyclization/Chlorination of Various Cyclic 8-Aryl-5-aza-5-tosyl-2-en-7-yn-1-ols

entry	substrate	$R_1$	$R_2$	product	yield (%)
1	1a	phenyl	Н	2a	83
2	1b	4-methylphenyl	Η	<b>2b</b>	80
3	1c	4-nitrophenyl	Η	2c	83
4	1d	3-carbethoxyphenyl	Η	2d	89
5	1e	4-phenylphenyl	H	2e	89
6	<b>1f</b>	4-bromophenyl	Η	2f	90
7	1g	phenyl	$CH_3$	$2\mathbf{g}$	97
8	1h	4-methoxyphenyl	H		
9	1i	H	H		

Scheme 3. FeCl<sub>3</sub>-Promoted Cylization/Chlorination of 10

a: n = 1, Ar = phenyl 86%
b: n = 1, Ar = 4-phenylphenyl 58%
c: n = 1, Ar = 4-methylphenyl 63%
d: n = 1, Ar = 4-bromophenyl 76%
e: n = 1, Ar = 3-carbethoxyphenyl 86%

**f**: n = 0, Ar = phenyl 81%

Scheme 1. Plausible Mechanism